

EVALUATION OF HARDNESS AND CORROSION BEHAVIOUR OF Al 7075/TiB₂/Cr₂O₃ COMPOSITE

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ABSTRACT

Aluminium composite is an attractive material for using in corrosive environment. Many studies have been conducted on corrosion rate in Aluminium composites. In this work Aluminium 7075 composite is reinforced with TiB₂ and Cr₂O₃ with different weight percentage of reinforcements. From the results it is understood that the density of the composite increases with more addition of reinforcements to the matrix. The uniform immersion test (ASTM-G31) in 90 ml of HCl was conducted. It was found that there is an increase in corrosion resistance when the reinforcements were added.

KEYWORDS: *Aluminium Composite, Uniform Immersion Test & Corrosion*

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INTRODUCTION

Aluminium metal is a popular material in many of the engineering applications like Aerospace application, Automobile applications etc. Nowadays the Aluminium alloys are slowly replaced with composite materials. The composite materials shows an improved material properties when compared to Aluminium alloys.

A study made by Gu et al. [13] shows that when Aluminium is reinforced with carbon fibers its corrosion rate decreases in distilled water. Bodunrin et al. [6] reported that Aluminium when reinforced with SiC, Gr and Al₂O₃ shows a decrease in corrosion rate when compared with parent alloy. A study made by Cheng et al. [8] shows that the addition of SiC in Aluminium 7075 alloy increases the corrosion rate. Bobić et al. [5] reported that the presence of Mica and Alumina in the matrix elevate the corrosion resistance of 356 alloy. Bobić et al. [4] made a study that the addition of graphite, glass and zircon particles improves the corrosion resistance in Zinc alloy metal matrix composite. Coleman et al. [9] reported that the corrosion rate for Al-7Si-0.5Mg alloy can be reduced with the addition of carbon fiber. El-Aziz et al. [10] reported that the presence of Al₂O₃ particle will decrease the corrosion rate in Al-Si alloy material composites. Abbass et al. [1] made a study on Aluminium 6061 composite and found that the addition of SiC will reduce the corrosion resistance rate significantly. Ahmad et al. [2] made a comparative study and found that corrosion rate of MMC's decreases with exposure in immersion test and pitting potential decreased with increase in volume fraction of SiC composites. Hihara and Latanision [15] reported that the presence of graphite fiber, SiC and TiB₂ improve the corrosion resistance in Aluminium 6061 metal matrix composites. A study made by Wielage et al. [19] shows that the corrosion rate can be improved further by adding diamond like carbon coated carbon fiber in Aluminium metal matrix composites. Cai et al. [7] reported that carbon fiber can increase the corrosion resistance in Aluminium composites when tested in artificial sea water. Sherif et al. [18] made a study and conclude that the corrosion rate increases with the addition of Graphite in pure

Aluminium composites. Saravanan et al. [17] reported that the addition of cenosphere will reduce corrosion rate for Aluminium 6063 metal matrix composites. Falcon et al. [12] made a study that the presence of TiC particles in Mg-Aluminium alloy composites will increase the corrosion resistance. Elcicek et al. [11] reported that addition of AlB₂ in Al-Cu alloy metal matrix will reduce the corrosion rate of the composites. The presence of α -Al amorphous structures in Aluminium matrix surface will reduce the corrosion rate which was reported by [16]. When Silicon Carbide particulates were used as reinforcements in 5A06 Aluminium matrix composite the overall corrosion rate decreases. This study was made by Aziz and Zhang [3]. The corrosion rate for Aluminium composite can be reduced by adding B₄C as reinforcement. This study was made by Han et al. [14]. When red mud was added to Aluminium 6061 metal matrix its overall corrosion rate decrease which was reported by Zucchi et al. [20].

From the above literatures it is clear that much work is not done on corrosion study in Al7075-TiB₂, Cr₂O₃ composites. In this present work evaluation of corrosion rate, strength and density have been carried out. Stir Casting method was adopted for fabricating the composite since it is an efficient manufacturing method for preparing composites.

EXPERIMENTAL METHOD

Stir casting method was selected for manufacturing composite. It is an efficient and economical method of manufacturing composites. The density of Aluminium 7075 is 2.65 g/cc. The density of TiB₂ and Cr₂O₃ are 4.52 g/cc and 5.22 g/cc respectively. Three hybrid composites were prepared. The weight percentage of TiB₂ selected were 10 % and maintained as constant. The weight percentage of Cr₂O₃ was varied as 4%, 6% and 8%.

Table 1: Weight Percentage of Reinforcements

Samples	TiB ₂ Weight %	Cr ₂ O ₃ Weight %
Sample 1	10 %	4 %
Sample 2	10 %	6 %
Sample 3	10 %	8 %

The Aluminium alloy was melted when it was heated to a temperature of 440°C. At this temperature there is a chance of oxidation. So in order to prevent oxidation degassing was done using nitrogen for a time period of 4 minutes. The reinforcements TiB₂ and Cr₂O₃ were also preheated. The Aluminium 7075 was again heated to 900°C. At this temperature stirring was done at 250-450 rpm. The stirring was continued for approximately 2 minutes and during this time the preheated reinforcements were added to the liquid metal. The stirring was continued for another 2 minutes. This was done to ensure uniform distribution of reinforcements in the material. The molten metal was poured to the prepared metallic mould. After pouring it was allowed to solidify at room temperature. After solidification the workpiece of required dimension was removed from the metallic mould.

SEM IMAGE ANALYSIS

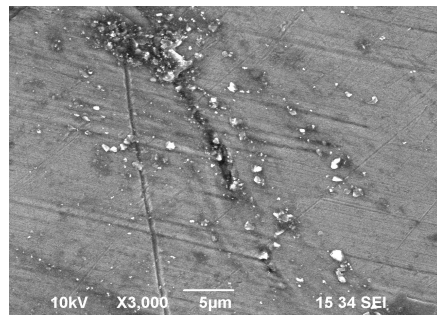


Figure 1: SEM Image of Sample 1

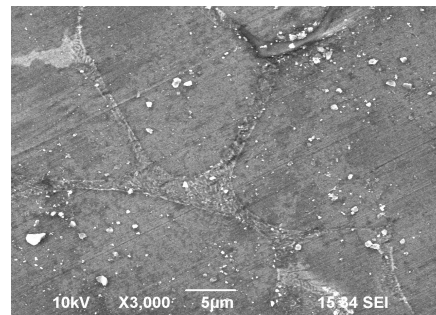


Figure 2: SEM Image of Sample 2

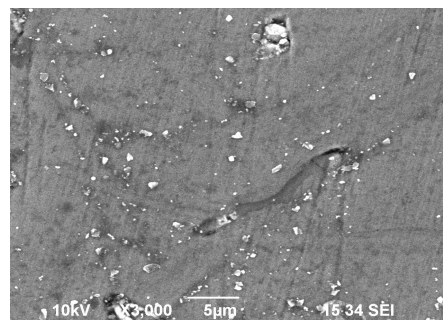


Figure 3: SEM Image of Sample 3

From the SEM image it can be clearly seen that the particles are uniformly distributed in Aluminium matrix. The presence of the materials can be ensured from EDX pattern. The images shows that the particles added during the manufacturing are present in the matrix. As the weight percentage of Cr₂O₃ was more in specimen 2 the EDX pattern was slightly varied. Similar variation can be seen in EDX pattern for specimen 3 also. From the below figures it is clear that good mechanical bonding was achieved between the matrix and reinforcements. Since mixing of reinforcements was made on semi solid state of Aluminium, it ensures the proper mixing of reinforcements in Aluminium matrix. TiB₂ and Cr₂O₃ provide sufficient wettability with present matrix and this ensures good mechanical bonding with the matrix. In figure 1 only lesser particles can be see. But in figure 2 and figure 3 more paticles can be seen. This is due to addition of more quantity of Cr₂O₃ particles in the Aluminium matrix. Since stirring was continued even after mixing reinforcements which ensures satisfactory distribution of reinforcements in Aluminium matrix.

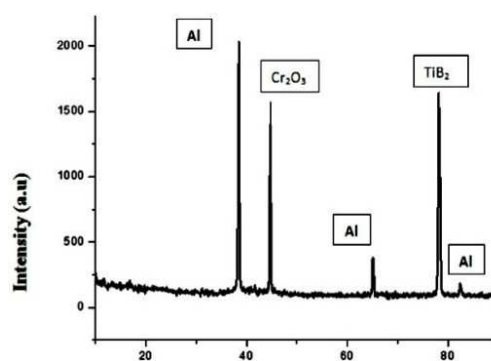


Figure 4: XRD image of Specimen 1

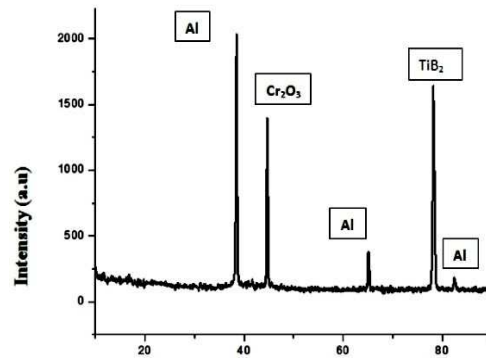


Figure 5: XRD Image of Specimen 2

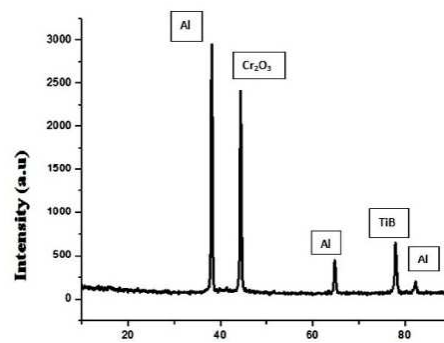


Figure 6: XRD image of Specimen 3

HARDNESS TESTING

For checking Vickers Hardness ASTM E10 standard was followed. The testing was conducted and hardness value was noted. The results show that the hardness value was significantly increased when percentage of Cr₂O₃ in Aluminium matrix was increased. The density of the specimens was also calculated. From the results it was clear that the density of the composites was higher than the parent Aluminium alloy. The density of the composite increases with increase in weight percentage of Cr₂O₃. Thus increase in density is due to the fact that more reinforced were filled in specimen porosity, thus further increase in weight of the component.

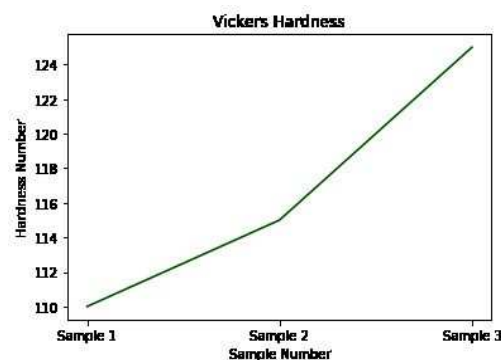


Figure 7: Hardness Test Values

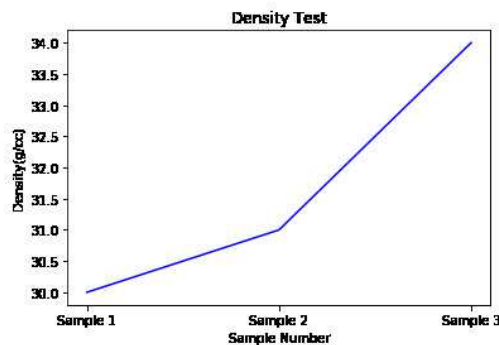


Figure 8: Density Test Values

CORROSION ANALYSIS

For analysing corrosion rate, the specimens are immersed in diluted 90 ml HCl solution. HCl was preferred over NaCl solution because the rate of corrosion will be more in HCl solution when compared to NaCl solution. Results show that there is a decrease in corrosion rate in the hybrid composites. From figure 9 it can be observed that the corrosion rate gradually decreases with the increase in weight percentage of Cr₂O₃. As the weight percentage of reinforcements were increased more pores in the parent material were filled with reinforcements. This in turn results in a decrease on the corrosion rate. Since more voids were filled this reduces lesser places for corrosion to start in the composite. The calculation for corrosion rate in mm/yr is as follows.

$$c = \frac{87.6 \times W}{\rho \times A \times t} \quad \rho = \text{density in g/cm}^3$$

W = loss of weight in mg

A = Area in cm²

t = time period in hours

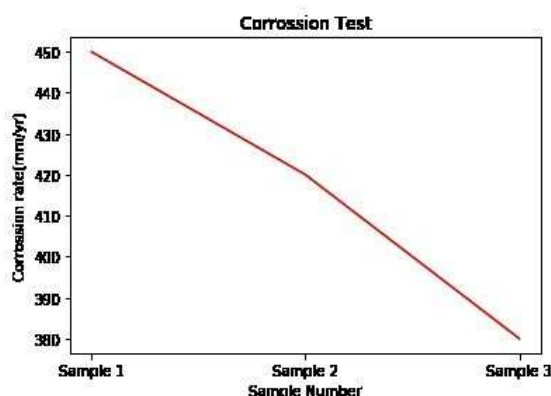


Figure 9: Corrosion Test Values

There are many factors that influence corrosion rate. Some of the factors are reinforcement shape, type, size, processing techniques and environmental factors. If the processing techniques for manufacturing composite materials are changed this can change the corrosion rate. There should be proper bonding between matrix and reinforcements. The strong bonding of matrix with reinforcement can lead to higher corrosion resistance. The increase in corrosion resistance

may also due to good bonding between parent material and reinforcements. If more reinforcement particles are bonded with parent material lesser will be corrosion rate. Corrosion rate decreases with weight percentage of reinforcements.

CONCLUSIONS

The present work can be concluded as follows

- The metal matrix hybrid composites were fabricated successfully using stirr casting technique.
- SEM image shows the dispersion of reinforcements in matrix phase were fairly uniform.
- The hardness of the composite increase when more Cr_2O_3 is added to the matrix.
- The corrosion resistance increase with increase in weight percentage of Cr_2O_3 .

REFERENCES

1. Muna K Abbass, Khairia S Hassan, and Abbas S Alwan. Study of corrosion resistance of aluminum alloy 6061/sic composites in 3.5% nacl solution. *International Journal of Materials, Mechanics and Manufacturing*, 3(1):31–35, 2015.
2. Zaki Ahmad, Amir Farzaneh, and BJ Abdul Aleem. Corrosion behavior of aluminium metal matrix composite. In *Recent Trends in Processing and Degradation of Aluminium Alloys*. InTech, 2011.
3. Irfan Aziz and Qi Zhang. Evaluation of pitting corrosion of the sic p/5 a 06 aluminum metal composite in 3.5% nacl solution by means of electrochemical impedance spectroscopy (eis). *International Journal of Modern Physics B*, 23(06n07):1497–1502, 2009.
4. B Bobić, S Mitrović, M Babić, and I Bobić. Corrosion of aluminium and zinc-aluminium alloys based metal-matrix composites. *Tribology in industry*, 31(3-4):44–53, 2009.
5. B Bobić, S Mitrović, M Babić, and I Bobić. Corrosion of metal-matrix composites with aluminium alloy substrate. *Tribology in industry*, 32(1):3–11, 2010.
6. Michael Oluwatosin Bodunrin, Kenneth Kanayo Alaneme, and Lesley Heath Chown. Aluminium matrix hybrid composites: a review of reinforcement philosophies; mechanical, corrosion and tribological characteristics. *journal of materials research and technology*, 4(4):434–445, 2015.
7. BP Cai, YH Liu, CK Ren, ZK Liu, XJ Tian, and ABB Abulimiti. Experimental study of galvanic corrosion behaviour of carbon fibre composite coupled to aluminium in artificial seawater. *Corrosion Engineering, Science and Technology*, 47(4):289–296, 2012.
8. YL Cheng, ZH Chen, HL Wu, and HM Wang. The corrosion behaviour of the aluminum alloy 7075/sicp metal matrix composite prepared by spray deposition. *Materials and Corrosion*, 58(4): 280–284, 2007.
9. SL Coleman, VD Scott, and B McEnaney. Corrosion behaviour of aluminium-based metal matrix composites. *Journal of materials science*, 29(11):2826–2834, 1994.
10. Khudhair, M. R., & Mallarapu, M. G. K. Frequency Responses Of Aluminum A356 Based On High Strength Alloy Composite (Hsap).
11. Khalid Abd El-Aziz, Dalia Saber, and Hossam El-Din M Sallam. Wear and corrosion behavior of al–si matrix composite reinforced with alumina. *Journal of Bio-and Tribo-Corrosion*, 1(1):5, 2015.

12. H Elcicek, Ö Savaş, Z Aydın, OK Özdemir, and R Kayikci. Corrosion behavior of in-situ al₂/al-cu metal matrix composite. *Acta Physica Polonica, A.*, 129(4), 2016.
13. LA Falcon, B Bedolla, J Lemus, C Leon, I Rosales, JG Gonzalez-Rodriguez, et al. Corrosion behavior of mg-al/tic composites in nacl solution. *International Journal of Corrosion*, 2011, 2011.
14. J Gu, GH Han, QW Sui, and CK Yao. Corrosion behaviour of a carbon aluminium composite. *Composites science and technology*, 31(2):111–120, 1988.
15. Yu-Mei Han, X Chen, et al. Electrochemical behavior of al-b₄c metal matrix composites in nacl solution. *Materials*, 8(9):6455–6470, 2015.
16. LH Hihara and RM Latanision. Galvanic corrosion of aluminum-matrix composites. *Corrosion*, 48(7):546–552, 1992.
17. Peng Liu, Qing-zhou Sun, Yan Liu, and Qi-lei Sun. Microstructure and corrosion properties of 5a06 aluminum matrix surface composite fabricated by friction stir processing. *Science and Engineering of Composite Materials*, 20(2):123–127, 2013.
18. V Saravanan, PR Thyla, N Nirmal, and SR Balakrishnan. Corrosion behavior of cenosphere - aluminium metal matrix composite in seawater condition. *International Journal of Chem Tech Research*, 8(2):726–731, 2015.
19. El-Sayed M Sherif, AA Almajid, Fahamsyah Hamdan Latif, and Harri Junaedi. Effects of graphite on the corrosion behavior of aluminum-graphite composite in sodium chloride solutions. *Int. J. Electrochem. Sci*, 6:1085–1099, 2011.
20. B Wielage, A Dorner, C Shürer, and JH Kim. Corrosion protection of carbon fibre reinforced aluminium composite by diamondlike carbon coatings. *Materials science and technology*, 16(3): 344–348, 2000.
21. F Zucchi, V Grassi, C Monticelli, and A Frignani. Corrosion behaviour of aluminium alloy metal matrix composite joints obtained by fsw. *Metallurgical Science and Tecnology*, 23(2), 2005.

